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Unit 6 -Chemical Bonding Test Review

1. The OCTET RULE states that atoms try to fill up their _____ and _____ orbitals in order to attain 8 valence electrons.
2. When drawing Lewis Dot structures, what do the dots around the symbol represent?
3. Draw Lewis Dot structures for the following compounds:
 - a. CO_2
 - b. MgCl_2
 - c. CH_4
4. Fill in the T-Chart below by completing the definition of each type of bond and providing 2 examples for each.

	<u>Ionic</u>	<u>Covalent</u>
Definition	A bond between elements from _____ sides of the periodic table; a bond between elements with very _____ electronegativities.	A bond between elements _____ each other on the periodic table; a bond between elements with _____ electronegativities.
Examples:		

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CH Cherryholmes

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