

Test bank chapter (9)

Choose the most correct answer:

1. The two types of chemical bonds commonly found in compounds are:
 - a) ionic and covalent.
 - b) ionic and electrolytic.
 - c) **ionic and covalent.**
 - d) electrolytic and compound.
2. The electrons used by atoms to form chemical bonds are the:
 - a) core electrons.
 - b) **valence electrons.**
 - c) lone pair electrons.
 - d) unpaired electrons.
3. "Atoms tend to gain, lose, or share electrons until they are surrounded by eight valence electrons" is a statement of:
 - a) the rule of octaves.
 - b) the double quartet rule.
 - c) the eight electron rule.
 - d) **the octet rule.**
4. When a transition metal atom becomes a +1 ion, the electron lost usually comes from what type of orbital?
 - a) p
 - b) f
 - c) d
 - d) s
5. A molecule of CS_2 contains:
 - a) two single bonds.
 - b) **two double bonds.**
 - c) one single bond and one double bond.
 - d) one single bond and one triple bond.
6. An atom in the ground state has atomic number $Z=5$. Choose the correct electron-dot structure which represents this atom? **ANS: B**
(A) $-\ddot{\text{N}}-\ddot{\text{N}}-$
(B) $\ddot{\text{N}}-\ddot{\text{N}}-$
(C) $-\ddot{\text{N}}-\ddot{\text{N}}-$
(D) $=\ddot{\text{N}}-\ddot{\text{N}}-$

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Larson**

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