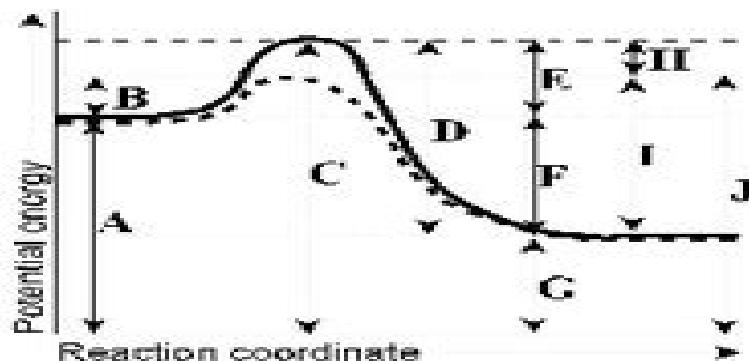


Interpreting Reaction Coordinates

The diagram below shows the reaction coordinate for a reversible catalyzed and uncatalyzed reaction. Referring to the diagram, answer the questions that follow.



<p>1. The reaction shown above is: (a) endothermic, (b) exothermic.</p> <p>2. Which lettered arrow represents the energy of the reactants for the forward reaction?</p> <p>3. Which lettered arrow represents the energy of the reactants for the reverse reaction?</p> <p>4. Which lettered arrow represents the energy of the products for the forward reaction?</p> <p>5. Which lettered arrow represents the energy of the products for the reverse reaction?</p> <p>6. Which lettered arrow represents ΔH for the forward catalyzed reaction?</p> <p>7. Which lettered arrow represents ΔH for the forward uncatalyzed reaction?</p> <p>8. Which lettered arrow represents ΔH for the reverse catalyzed reaction?</p> <p>9. Which lettered arrow represents ΔH for the reverse uncatalyzed reaction?</p>	<p>10. Which lettered arrow represents activation energy for the forward catalyzed reaction?</p> <p>11. Which lettered arrow represents activation energy for the forward uncatalyzed reaction?</p> <p>12. Which lettered arrow represents activation energy for the reverse catalyzed reaction?</p> <p>13. Which lettered arrow represents activation energy for the reverse uncatalyzed reaction?</p> <p>14. Which lettered arrow represents energy of the activated complex for the catalyzed reaction?</p> <p>15. Which lettered arrow represents energy of the activated complex for the uncatalyzed reaction?</p> <p>16. Which lettered arrow represents the difference between the activation energies of the catalyzed and uncatalyzed reactions?</p> <p>17. Which lettered arrow represents the difference between the energies of the activated complex for the catalyzed and uncatalyzed reactions?</p> <p>18. The reverse reaction is: (a) endothermic, (b) exothermic.</p>
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Interpreting Reaction Coordinates Answer Key

SJ Ball

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