

Ionic and Covalent Naming Practice **Answers**

For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately.

1)	Na_2CO_3	sodium carbonate
2)	P_4O_{10}	tetraphosphorus decaoxide
3)	N_2O_5	dinitrogen pentoxide
4)	FeSO_4	iron (II) sulfate
5)	SiO_2	silicon dioxide
6)	FeCl_3	iron (III) chloride
7)	CoBr_2	cobalt (III) bromide
8)	B_2H_6	diboron tetrahydride
9)	CO	carbon monoxide
10)	PCl_3	phosphorus trichloride

For each of the following questions, determine whether the compound is ionic or covalent and write the appropriate formula for it.

11)	dinitrogen trioxide	N_2O_3
12)	nitrogen dioxide	NO_2
13)	sulfur pentoxide	SO_5
14)	lithium acetate	$\text{LiC}_2\text{H}_3\text{O}_2$
15)	phosphorus trifluoride	PF_3
16)	vanadium (V) oxide	V_2O_5
17)	aluminum hydroxide	$\text{Al}(\text{OH})_3$
18)	zinc sulfide	ZnS
19)	silicon tetrafluoride	SiF_4
20)	silver phosphate	Ag_3PO_4

Ionic And Covalent Nomenclature Ws Answers

Holt, Rinehart and Winston Staff

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