

KEY

Ionic Formula Practice

Name	Cation	Anion	Formula
1. calcium bromide	Ca^{+2}	Br^{-1}	Ca Br_2
2. magnesium phosphide	Mg^{+2}	P^{-3}	$\text{Mg}_3 \text{P}_2$
3. aluminum oxide	Al^{+3}	O^{-2}	$\text{Al}_2 \text{O}_3$
4. barium sulfide	Ba^{+2}	S^{-2}	Ba S_2
5. sodium phosphide	Na^{+1}	P^{-3}	$\text{Na}_3 \text{P}$
6. strontium chloride	Sr^{+2}	Cl^{-1}	Sr Cl_2
7. calcium nitride	Ca^{+2}	N^{-3}	$\text{Ca}_3 \text{N}_2$
8. ammonium bromide	NH_4^{+1}	Br^{-1}	$\text{NH}_4 \text{Br}$
9. calcium sulfate	Ca^{+2}	SO_4^{-2}	Ca SO_4
10. magnesium nitrate	Mg^{+2}	NO_3^{-1}	$\text{Mg}(\text{NO}_3)_2$
11. sodium carbonate	Na^{+1}	CO_3^{-2}	$\text{Na}_2 \text{CO}_3$
12. zinc hydroxide	Zn^{+2}	OH^{-1}	$\text{Zn}(\text{OH})_2$
13. barium nitrite	Ba^{+2}	NO_2^{-1}	$\text{Ba}(\text{NO}_2)_2$
14. aluminum sulfite	Al^{+3}	SO_3^{-2}	$\text{Al}_2(\text{SO}_3)_3$
15. lead II nitrate	Pb^{+2}	NO_3^{-1}	$\text{Pb}(\text{NO}_3)_2$
16. copper I phosphate	Cu^{+1}	PO_4^{-3}	$\text{Cu}_3 \text{PO}_4$
17. ammonium carbonate	NH_4^{+1}	CO_3^{-2}	$(\text{NH}_4)_2 \text{CO}_3$
18. tin IV nitrate	Sn^{+4}	NO_3^{-1}	$\text{Sn}(\text{NO}_3)_4$
19. nickel II acetate	Ni^{+2}	$\text{C}_2\text{H}_3\text{O}_2^{-1}$	$\text{Ni}(\text{C}_2\text{H}_3\text{O}_2)_2$
20. chromium III sulfate	Cr^{+3}	SO_4^{-2}	$\text{Cr}_2(\text{SO}_4)_3$
21. iron III nitrite	Fe^{+3}	NO_2^{-1}	$\text{Fe}(\text{NO}_2)_3$
22. cobalt III hydroxide	Co^{+2}	OH^{-1}	$\text{Co}(\text{OH})_2$

Ionic Formula Writing Kit Lab Answer Key

Tracy Lynn Haroff

Ionic Formula Writing Kit Lab Answer Key:

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Laboratory Experiments for General Chemistry Harold R. Hunt,Toby F. Block,George M. McKelvy,1998 **Laboratory Manual** Jo A. Beran,1990 **Improving Student Comprehension in Chemistry Laboratories** Tracy Lynn Haroff,2006

Biology/science Materials Carolina Biological Supply Company,1991 General, Organic, and Biological Chemistry Study Guide and Selected Solutions Karen C. Timberlake,2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems *Scientific and Technical Aerospace Reports* ,1980 Community and Junior College Journal ,1974 *Multiscale Operational Organic Chemistry* John W. Lehman,2009 This comprehensive laboratory text provides a thorough introduction to all of the significant operations used in the organic lab and includes a large selection of traditional scale and microscale experiments and minilabs Its unique problem solving approach encourages students to think in the laboratory by solving a scientific problem in the process of carrying out each experiment The Second Edition contains a new introductory section Chemistry and the Environment which includes a discussion of the principles of green chemistry Several green experiments have been added and some experiments from the previous editions have been revised to make them greener **Aero/space Engineering** ,1962 **INIS Atomindeks** ,1988 **Druggists' Circular and Chemical Gazette** ,1905 Includes Red book price list section title varies slightly issued semiannually 1897 1906

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