

Name _____ Date _____ Class _____

CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

Matter—Properties and Changes**Section 3.1 Properties of Matter**

Fill in the blank, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical density	mass properties	physical substances
---------------------	--------------------	------------------------

Matter is anything with **mass** and **volume**.A **substance** is a form of matter with a uniform and unchanging composition.Substances have specific, unchanged **properties** that can be observed.Substances have both **physical** and **chemical** properties. **Physical**properties can be observed without changing a substance's chemical composition. Color, hardness, and **density**, for example. Other properties cannot be

observed without changing the composition of a substance. These are called

chemical properties. An example is the tendency of zinc to form
rust when exposed to air.Label each property as either **physical** or **chemical**.physical 1. Chemical formula H_2O

chemical 2. Turns green when exposed to air

chemical 3. Remains unchanged when in the presence of nitrogen

physical 4. Colorless

physical 5. Solid at normal temperatures and pressures

chemical 6. Ability to combine with another substance

physical 7. Melting point

physical 8. Liquid at normal temperatures and pressures

physical 9. Boiling point is 100°C

physical 10. Conducts electricity

physical 11. Density is $\frac{10}{100}$

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CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

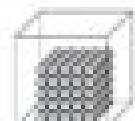
Section 3.1 continued

In your textbook, read about states of matter.

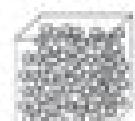
Label each drawing with one of these words: solid, liquid, gas.



gas



solid



liquid

For each statement below, write true or false.

true 11. All matter in everyday life exists in one of these physical forms.

true 12. A solid has definite shape and volume.

false 13. A liquid has a definite shape and takes on the volume of its container.

true 14. A gas has both the shape and the volume of its container.

false 15. The particles in a gas cannot be compressed into a smaller volume.

false 16. Liquids tend to contract when heated.

false 17. The particles in a solid are spaced far apart.

false 18. The words gas and vapor can be used interchangeably.

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Peter Scott,Rachael Enriquez Porter

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