

## Section A

For each question there are four possible answers **A**, **B**, **C** and **D**. Choose the one you consider to be correct.

Use of the Data Booklet may be appropriate for some questions.

1 Compound X is an organic compound that contains 30.6% carbon, 3.8% hydrogen, 20.4% oxygen and 45.2% chlorine by mass.

What is the empirical formula of X?

A  $C_2H_3OCl$       B  $C_2H_4OCl$       C  $C_3H_2OCl$       D  $C_4H_3O_2Cl_2$

2 A sample of propane,  $C_3H_8$ , with a mass of 9.61 g is completely combusted in an excess of oxygen under room conditions.

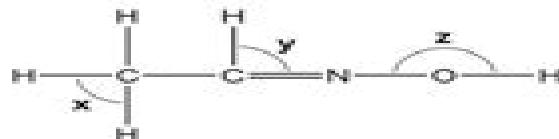
Which volume of carbon dioxide gas is produced?

A  $4.89 \text{ dm}^3$       B  $5.24 \text{ dm}^3$       C  $14.7 \text{ dm}^3$       D  $15.7 \text{ dm}^3$

3 Which atom has the same number of electrons as an ammonium ion?

A Mg      B Na      C Ne      D O

4 Ethanal reacts with hydroxylamine,  $NH_2OH$ , to form the molecule shown.



What is the order of increasing bond angle in this structure from smallest to largest?

A  $z, x, y$       B  $y, z, x$       C  $x, z, y$       D  $z, y, x$

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### **Table of Contents May 2013 9701 A Level Chemistry Paper**

1. Understanding the eBook May 2013 9701 A Level Chemistry Paper
  - The Rise of Digital Reading May 2013 9701 A Level Chemistry Paper
  - Advantages of eBooks Over Traditional Books
2. Identifying May 2013 9701 A Level Chemistry Paper
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an May 2013 9701 A Level Chemistry Paper
  - User-Friendly Interface
4. Exploring eBook Recommendations from May 2013 9701 A Level Chemistry Paper
  - Personalized Recommendations
  - May 2013 9701 A Level Chemistry Paper User Reviews and Ratings

- May 2013 9701 A Level Chemistry Paper and Bestseller Lists
- 5. Accessing May 2013 9701 A Level Chemistry Paper Free and Paid eBooks
  - May 2013 9701 A Level Chemistry Paper Public Domain eBooks
  - May 2013 9701 A Level Chemistry Paper eBook Subscription Services
  - May 2013 9701 A Level Chemistry Paper Budget-Friendly Options
- 6. Navigating May 2013 9701 A Level Chemistry Paper eBook Formats
  - ePUB, PDF, MOBI, and More
  - May 2013 9701 A Level Chemistry Paper Compatibility with Devices
  - May 2013 9701 A Level Chemistry Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of May 2013 9701 A Level Chemistry Paper
  - Highlighting and Note-Taking May 2013 9701 A Level Chemistry Paper
  - Interactive Elements May 2013 9701 A Level Chemistry Paper
- 8. Staying Engaged with May 2013 9701 A Level Chemistry Paper
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers May 2013 9701 A Level Chemistry Paper
- 9. Balancing eBooks and Physical Books May 2013 9701 A Level Chemistry Paper
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection May 2013 9701 A Level Chemistry Paper
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine May 2013 9701 A Level Chemistry Paper
  - Setting Reading Goals May 2013 9701 A Level Chemistry Paper
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of May 2013 9701 A Level Chemistry Paper
  - Fact-Checking eBook Content of May 2013 9701 A Level Chemistry Paper
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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