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Eugen Hausbrand



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PREFACE IN the present volume dealing with the Chemical Technology of the Textile Fibres except as concerns the dye stuffs which will be treated in a separate work the author has been obliged to condense the available matter as much as possible in order to preserve the form of a text book Nevertheless it seemed necessary in certain cases in the interests of the book to give definite data and an exact description of individual processes In such instances the details have been gathered exclusively either from the authors personal experience or from reliable sources The most important part of the book is the chapter treating of dyeing whilst on the other hand the subject of printing had to be dealt with in a more general fashion the materials being less suitable for treatment in text book style The author thinks it desirable to point out that in the present work an attempt has been made to completely separate the chemical and mechanical technology of the subject a standpoint he considers justified by the extensive area occupied by each of these branches Hence only a few sketches of apparatus have been given and the methods of dressing the finished goods have been described very briefly since they almost entirely belong to the domain of mechanical technology

GEOEG VON GEOEGIEVICS Artificial Fibres Mineral Vegetable Cellulose Cotton Bombax Cotton Vegetable Silk Flax Hemp Jute Ramie Rhea China Grass Nettle Fibre Contents include CHAPTER I THE TEXTILE FIBRES Distinguishing Tests for the Various Fibres Animal Fibres Silk Animal Hairs Sheeps Wool Goat Wool and Camel Wool Artificial Wool Wool Substitutes Conditioning CHAPTER II WASHING BLEACHING CARBONISING Washing and Bleaching Definition Bleaching Agents Cotton Bleaching PAGE iii 1 2 2 3 8 12 12 12 16 17 19 20 2 2 23 34 35 45 46 19 50 53 viii CONTENTS Linen Bleaching Ramie Bleaching Hemp Bleaching Jute Bleaching 76 Scouring and Bleaching Silk 77 Washing and Bleaching Wool 80 Blueing or White 86 Dyeing Carbonising 87 CHAPTER III MORDANTS AND MORDANTING Mordants 95 Mordanting Wool 96 Mordanting Silk 98 Mordanting Cotton 99 Alumina Mordants 102 Mordants Iron Mordants 106 Chrome 108 Tin Mordants 112 Copper and other Mordants 114 The Fixing Agents Acid Mordants 115 Tannic Acids Oleic Acids PAGE 116 122 CHAPTER IV DYEING 1 Theory of Colour Combination of Colours Dyeing to Pattern 125 2 Theory of Dyeing 130 3 Classification of Dye Stuffs Methods of Dyeing 138 Application of Acid Dye Stuffs Application of Basic v Dye Stuffs 143 Application of Direct or Substantive Cotton Dyes Dyes 146 Application of the Mordant 154 Dyeing with Cochineal 160 Dyeing with Catechu 178 Black and Blue Dyeings with Logwood on Wool 163 Turkey Red Dyeing 172 Black Dyeing Cotton with Logwood 180 **Bone Products and Manures** Thomas Lambert, 1901 **The art of dyeing wool, silk, and cotton**, tr.

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web how to balance $\text{hno}_3 + \text{mg} \rightarrow \text{mg}(\text{no}_3)_2 + \text{h}_2$ word equation nitric acid magnesium magnesium nitrate hydrogen gas type of chemical reaction for this reaction we have a single displacement reaction balancing strategies in this single displacement reaction the mg replaces the h in the hno_3

how to balance $\text{mg} + \text{hno}_3 \rightarrow \text{mg}(\text{no}_3)_2 + \text{no}_2 + \text{h}_2\text{o}$ magnesium youtube - May 08 2023

web nov 20 2020 in this video we ll balance the equation $\text{mg} + \text{hno}_3 \rightarrow \text{mg}(\text{no}_3)_2 + \text{no}_2 + \text{h}_2\text{o}$ and provide the correct coefficients for each compound to balance $\text{mg} + \text{hno}_3 \rightarrow \text{mg}(\text{no}_3)_2 + \text{no}_2 + \text{h}_2\text{o}$ you ll need to be

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web magnesium hydroxide nitric acid balanced total equation $\text{mg}(\text{oh})_2 + \text{aq} + 2 \text{hno}_3 \rightarrow \text{aq} + 2 \text{h}_2\text{o} + \text{mg}(\text{no}_3)_2$ total ionic equation $\text{mg}^{2+} + 2 \text{oh}^- + 2 \text{h}^+ + 2 \text{no}_3^- \rightarrow 2 \text{h}_2\text{o} + \text{mg}^{2+} + 2 \text{no}_3^-$ net ionic equation $2 \text{oh}^- + 2 \text{h}^+ \rightarrow 2 \text{h}_2\text{o}$ sulfuric acid and lithium hydroxide balanced total equation $\text{h}_2\text{so}_4 + \text{aq} + 2 \text{lioh} \rightarrow \text{aq} + 2 \text{h}_2\text{o}$