

Electron Configurations Worksheet - Answers

Write the complete ground state electron configurations for the following:

- | | | |
|----|----------|--|
| 1) | lithium | $1s^2 2s^1$ |
| 2) | oxygen | $1s^2 2s^2 2p^4$ |
| 3) | calcium | $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2$ |
| 4) | titanium | $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^2$ |
| 5) | rubidium | $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$ |
| 6) | lead | $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10} 6p^2$ |
| 7) | erbium | $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{12}$ |

Write the abbreviated ground state electron configurations for the following:

- | | | |
|-----|----------|---|
| 8) | helium | $1s^2$ (this one cannot be abbreviated) |
| 9) | nitrogen | $[\text{He}] 2s^2 2p^3$ |
| 10) | chlorine | $[\text{Ne}] 3s^2 3p^5$ |
| 11) | iron | $[\text{Ar}] 4s^2 3d^6$ |
| 12) | zinc | $[\text{Ar}] 4s^2 3d^{10}$ |
| 13) | barium | $[\text{Xe}] 6s^2$ |
| 14) | polonium | $[\text{Xe}] 6s^2 4f^{14} 5d^{10} 6p^4$ |

Electron Configuration Worksheet And More Key

Ralph Thomas Myers

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