

**R** : Conduction of water by vessel elements is an active process with energy supplied by xylem parenchyma rich in mitochondria

**13. A** : Polytene chromosomes have a high amount of DNA.  
**R** : Polytene chromosomes are formed by repeated replication of chromosomal DNA without separation of chromatids

**14. A** : UV radiation causes photo dissociation of ozone into  $O_2$  and O, thus causing damage to the stratospheric ozone layer.  
**R** : Ozone hole is resulting in global warming and climate change

**15. A** : The concentration of methane in the atmosphere has more than doubled in the last 250 years.  
**R** : Wetlands and rice fields are the major sources of methane

**16. A** : In tropical rain forests, O-Horizon and A-Horizon of soil profile are shallow and nutrient-poor.  
**R** : Excessive growth of micro-organisms in the soil depletes its organic content.

**17. A** : Gram-negative bacteria do not retain the stain when washed with alcohol.  
**R** : The outer face of the outer membrane of Gram-negative bacteria contains lipopolysaccharides, a part of which is integrated into the membrane lipids.

**18. A** : Under conditions of high intensity and limited  $CO_2$  supply, photo respiration has a useful role in protecting the plants from photo-oxidative damage.  
**R** : If enough  $CO_2$  is not available to utilize light energy for carboxylation to proceed, the excess energy may not cause damage to plants.

**19. A** : Photo synthetically  $C_4$  plants are less efficient than  $C_3$  plants.  
**R** : The operation of  $C_4$  pathway requires the involvement of only bundle-sheath cells.

**20. A** : Eukaryotic cells have the ability to adopt a variety of shapes and carry out directed movements.  
**R** : There are three principal types of protein filaments- microfilaments, microtubules and intermediate filaments, which constitute the cytoskeleton.

**21. A** : In the iodometric titration starch is used as an indicator.  
**R** : Starch is polysaccharide

**22. A** : Molecular nitrogen is less reactive than molecular oxygen.  
**R** : The bond length of  $N_2$  is shorter than that of oxygen.

**23. A** :  $[Co(NO_2)_3(NH_3)_3]$  does not show optical isomerism.  
**R** : It has a plane of symmetry.

**24. A** :  $E^\circ$  for  $Mn^{3+}/Mn^{2+}$  is more positive than  $Cr^{3+}/Cr^{2+}$ .

**R** : The third ionization energy of Mn is larger than that of Cr

**25. A** :  $K_2Cr_2O_7$  is used as a primary standard in volumetric analysis.  
**R** : It has a good solubility in water

**26. A** : Silicones are hydrophobic in nature.  
**R** : Si - O - Si linkages are moisture sensitive

**27. A** : According to transition state theory, for the formation of an activated complex, one of the vibrational degrees of freedom is converted into a translational degree of freedom.  
**R** : Energy of the activated complex is higher than the energy of reactant molecules

**28. A** : Water in liquid state is more stable than ice at room temperature.  
**R** : Water in liquid form has higher entropy than ice.

**29. A** :  $Sb_2S_3$  is not soluble in yellow ammonium sulphide.  
**R** : The common ion effect due to  $S^{2-}$  ions reduces the solubility of  $Sb_2S_3$ .

**30. A** : Graphite is an example of tetragonal crystal system.  
**R** : For a tetragonal system  
 $a = b \neq c$ ,  $\alpha = \beta = 90^\circ$ ,  $\gamma = 120^\circ$

**31. A** : For the Daniel cell  $Zn|Zn^{2+}||Cu^{2+}|Cu$  with  $E_{cell} = 1.1$  V, the application of opposite potential greater than 1.1 V results into flow of electron from cathode to anode.  
**R** : Zn is deposited at anode and Cu is dissolved at cathode

**32. A** :  $Fe^{2+}$  can be used for coagulation of  $As_2S_3$  sol.  
**R** :  $Fe^{3+}$  reacts with  $As_2S_3$  to give  $Fe_2S_3$ .

**33. A** : If red blood cells were removed from the body and placed in pure water, pressure inside the cells increases.  
**R** : The concentration of salt content in the cells increases.

**34. A** : Change in colour of acidic solution of potassium dichromate by breath is used to test drunk drivers.  
**R** : Change in colour is due to the complication of alcohol with potassium dichromate

**35. A** : Anilinium chloride is more acidic than ammonium chloride.  
**R** : Anilinium ion is resonance-stabilized

**36. A** : Diastereoisomers have different physical properties.  
**R** : They are non-superimposable mirror images.

**37. A** : The presence of nitro group facilitates nucleophilic substitution reactions in aryl halides.  
**R** : The intermediate carbonium is stabilized due to the presence of nitro group

**38. A** : 1, 3-Butadiene is the monomer for natural rubber.

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web microeconomics section i time 70 minutes 60 questions directions each of the question or incomplete statements below is followed by five suggested answers or completions select the one that is best in each case and then fill in the corresponding oval on the answer sheet 1 which of the following constitute the

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web c for a firm selling in a perfectly competitive market the marginal revenue product mrp is equal to the marginal physical product mpp times the market price here 5 units per labor input times 10 per unit of output equals 50 choice c is the correct answer while the other choices do not follow the

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web ap microeconomics 2007 scoring guidelines question 1 12 points 4 3 3 2 a 4 points one point is earned for correctly

labeled axes and an  $MR$  curve below a downward sloping demand curve one point is earned for showing profit maximizing  $q$  at  $MC = MR$  one point is earned for identifying  $P$  on the demand curve above  $q$

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