

**R** : Conduction of water by vessel elements is an active process with energy supplied by xylem parenchyma rich in mitochondria

13. **A** : Polytene chromosomes have a high amount of DNA

**R** : Polytene chromosomes are formed by repeated replication of chromosomal DNA without separation of chromatids

14. **A** : UV radiation causes photo dissociation of ozone into  $O_2$  and  $O$ , thus causing damage to the stratospheric ozone layer.

**R** : Ozone hole is resulting in global warming and climate change

15. **A** : The concentration of methane in the atmosphere has more than doubled in the last 250 years.

**R** : Wetlands and rice fields are the major sources of methane

16. **A** : In tropical rain forests, O-Horizon and A-Horizon of soil profile are shallow and nutrient-poor.

**R** : Excessive growth of micro-organisms in the soil depletes its organic content.

17. **A** : Gram-negative bacteria do not retain the stain when washed with alcohol

**R** : The outer face of the outer membrane of Gram-negative bacteria contains lipopolysaccharides, a part of which is integrated into the membrane lipids.

18. **A** : Under conditions of high intensity and limited  $CO_2$  supply, photo respiration has a useful role in protecting the plants from photo-oxidative damage.

**R** : If enough  $CO_2$  is not available to utilize light energy for carboxylation to proceed, the excess energy may not cause damage to plants

19. **A** : Photo synthetically  $C_4$  plants are less efficient than  $C_3$  plants

**R** : The operation of  $C_4$  pathway requires the involvement of only bundle-sheath cells.

20. **A** : Eukaryotic cells have the ability to adopt a variety of shapes and carry out directed movements

**R** : There are three principal types of protein filaments- microfilaments, microtubules and intermediate filaments, which constitute the cytoskeleton.

21. **A** : In the iodometric titration starch is used as an indicator

**R** : Starch is polysaccharide

22. **A** : Molecular nitrogen is less reactive than molecular oxygen

**R** : The bond length of  $N_2$  is shorter than that of oxygen

23. **A** :  $[Co(NO_2)_3(NH_3)_3]$  does not show optical isomerism

**R** : It has a plane of symmetry.

24. **A** :  $E^\circ$  for  $Mn^{3+}/Mn^{2+}$  is more positive than  $Cr^{3+}/Cr^{2+}$

**R** : The third ionization energy of Mn is larger than that of Cr

25. **A** :  $K_2Cr_2O_7$  is used as a primary standard in volumetric analysis

**R** : It has a good solubility in water

26. **A** : Silicones are hydrophobic in nature

**R** : Si - O - Si linkages are moisture sensitive

27. **A** : According to transition state theory, for the formation of an activated complex, one of the vibrational degrees of freedom is converted into a translational degree of freedom

**R** : Energy of the activated complex is higher than the energy of reactant molecules

28. **A** : Water in liquid state is more stable than ice at room temperature

**R** : Water in liquid form has higher entropy than ice.

29. **A** :  $Sb_2S_3$  is not soluble in yellow ammonium sulphide

**R** : The common ion effect due to  $S^{2-}$  ions reduces the solubility of  $Sb_2S_3$

30. **A** : Graphite is an example of tetragonal crystal system

**R** : For a tetragonal system

$a = b \neq c$ ,  $\alpha = \beta = 90^\circ$ ,  $\gamma = 120^\circ$

31. **A** : For the Daniel cell  $Zn|Zn^{2+}||Cu^{2+}|Cu$  with  $E_{cell} = 1.1$  V, the application of opposite potential greater than 1.1 V results into flow of electron from cathode to anode

**R** : Zn is deposited at anode and Cu is dissolved at cathode

32. **A** :  $Fe^{3+}$  can be used for coagulation of  $As_2S_3$  sol.

**R** :  $Fe^{3+}$  reacts with  $As_2S_3$  to give  $Fe_2S_3$

33. **A** : If red blood cells were removed from the body and placed in pure water, pressure inside the cells increases

**R** : The concentration of salt content in the cells increases

34. **A** : Change in colour of acidic solution of potassium dichromate by breath is used to test drunk drivers.

**R** : Change in colour is due to the complication of alcohol with potassium dichromate

35. **A** : Anilinium chloride is more acidic than ammonium chloride

**R** : Anilinium ion is resonance-stabilized

36. **A** : Diastereoisomers have different physical properties

**R** : They are non-superimposable mirror images.

37. **A** : The presence of nitro group facilitates nucleophilic substitution reactions in aryl halides

**R** : The intermediate carbanion is stabilized due to the presence of nitro group

38. **A** : 1, 3-Butadiene is the monomer for natural rubber

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web microeconomics section i time 70 minutes 60 questions directions each of the question or incomplete statements below is followed by five suggested answers or completions select the one that is best in each case and then fill in the corresponding oval on the answer sheet 1 which of the following constitute the

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web c for a firm selling in a perfectly competitive market the marginal revenue product m<sub>rp</sub> is equal to the marginal physical product m<sub>pp</sub> times the market price here 5 units per labor input times 10 per unit of output equals 50 choice c is the correct answer while the other choices do not follow the

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web exam format the ap microeconomics exam includes two sections the first section contains 60 multiple choice questions the second section is the free response section which includes one long question and two short questions below is a detailed breakdown of both the multiple choice and free response sections

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web ap microeconomics 2007 scoring guidelines question 1 12 points 4 3 3 2 a 4 points one point is earned for correctly

labeled axes and an  $MR$  curve below a downward sloping demand curve one point is earned for showing profit maximizing  $Q$  at  $MC = MR$  one point is earned for identifying  $P$  on the demand curve above  $Q$

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web key details and format of the exam the 2007 ap microeconomics exam is a comprehensive test that assesses students understanding of economic concepts and their ability to apply them to real world scenarios this exam consists of two sections multiple choice and free response each section is designed to test different skills and knowledge