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web nov 20 2020 in this video we ll balance the equation $\text{Mg} + \text{HNO}_3 \rightarrow \text{Mg(NO}_3)_2 + \text{H}_2 + \text{H}_2\text{O}$ and provide the correct coefficients for each compound to balance $\text{Mg} + \text{HNO}_3 \rightarrow \text{Mg(NO}_3)_2 + \text{H}_2 + \text{H}_2\text{O}$ you ll need to be

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web magnesium hydroxide nitric acid balanced total equation $\text{Mg(OH)}_2 + 2\text{HNO}_3 \rightarrow \text{Mg(NO}_3)_2 + 2\text{H}_2\text{O}$ mg NO_3^- aq total ionic equation $\text{Mg}^{2+} + 2\text{OH}^- + 2\text{H}^+ + 2\text{NO}_3^- \rightarrow \text{Mg}^{2+} + 2\text{NO}_3^- + 2\text{H}_2\text{O}$ net ionic equation $2\text{OH}^- + 2\text{H}^+ \rightarrow 2\text{H}_2\text{O}$ sulfuric acid and lithium hydroxide balanced total equation $\text{H}_2\text{SO}_4 + 2\text{LiOH} \rightarrow \text{Li}_2\text{SO}_4 + 2\text{H}_2\text{O}$

