

## Multiple Choice Reaction Mechanisms

23. Given the following reaction mechanism, what is the equation for the overall reaction?



A) 2A  $\rightarrow$  2E + F      B) 2A + B + 2C  $\rightarrow$  D + 2E + F      C) 2A + 2C  $\rightarrow$  2E + F  
D) 2A + B + 2C + D  $\rightarrow$  B + 2C + D + 2E + F      E) 2A + C  $\rightarrow$  2E + F

ANS: A

11. Which step below is the rate-determining step?

1)	W + XY $\rightarrow$ WX + Y	fast
2)	W + WX $\rightarrow$ W <sub>2</sub> X	very fast
3)	W <sub>2</sub> X + Z $\rightarrow$ WXZ + W	slow
4)	WXZ + Z $\rightarrow$ W <sub>2</sub> XZ <sub>2</sub>	fast

A) 1      B) 2      C) 3      D) 4

ANS: C

12. Which of the following rate law expressions would support the mechanism below?

1)	W + XY $\rightarrow$ WX + Y	fast
2)	W + WX $\rightarrow$ W <sub>2</sub> X	very fast
3)	W <sub>2</sub> X + Z $\rightarrow$ WXZ + W	slow
4)	WXZ + Z $\rightarrow$ W <sub>2</sub> XZ <sub>2</sub>	fast

A)  $r = k[W_2X][Z]$       B)  $r = k[W]^2$       C)  $r = k[WXZ][Z]$       D)  $r = k[W]^2[XY]$

ANS: A

13. Consider the following reaction mechanism:

1)	A $\rightarrow$ D + E	SLOW
2)	D + B $\rightarrow$ C	FAST

The overall reaction is:



How might the rate of the forward reaction be increased?

A) increase [A]      B) increase [B]      C) decrease temperature      D) both A and B

ANS: A

4. Consider the following reaction mechanism:



In this reaction, H<sub>2</sub> is a:

# Modern Chemistry Reaction Energy Multiple Choice Answers

**Lawrence F. Tuleen, Willard L.  
Muehl, George Sherman Porter**

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