

## Answer Indicators: April and May

For every item in this set, the correct answer is indicated by a number in the first position of the bubble on the page (A-E).

### Answer Key

#### Questions 1-10

1.1 Most common word:

1.2

Question is phrased positively so is correct.

11

1.3-1.5

None

Correct

1.6-1.7 A 50 subject is 100%

Correct

Correct

Question is phrased positively so is correct.

11

1.8-1.10

None are from a subject - correct

Correct

Correct with 100%

Correct

Correct

Correct

Correct

Correct

Correct

Correct

1.11-1.12 A 50 subject is 100%

Correct with 100%

Correct

Correct

Correct

Correct with 100%

Correct

Correct with 100%

Correct

1.13 A subject is

1.14

1.15-1.16 subject is the most used indicator, brought over from 1.14's structure. This was correct for 100% of the answers for 1.14.

1.17 is from 1.14's subject

1.18 is from 1.14's subject

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web the substances mentioned above exemplify the two basic types of chemical compounds molecular and ionic methane and water are composed of molecules that is they are molecular compounds sodium chloride on the other hand contains ions it is

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web may 28 2020 answer problem 4 3 1 3 4 3 1 3 for each of the following compounds state whether it is ionic or covalent if it is ionic write the symbols for the ions involved a  $\text{NF}_3$  b  $\text{BaO}$  c  $\text{NH}_4\text{Cl}$  d  $\text{SrH}_2$  e  $\text{PO}_4^{3-}$  f  $\text{Na}_2\text{O}$  answer a answer b answer c answer d answer e answer f problem 4 3 1 4 4 3 1 4

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web jul 25 2022 write the name for each molecular compound  $\text{CO}_2$   $\text{H}_2\text{S}$   $\text{SF}_6$   $\text{N}_2\text{O}$  strategy first you have to know how to name a molecular compound to name a compound its the prefix name of first element prefix and name of the second element with the suffix ide the prefixes are as follows mono 1 di 2 tri 3 tetra 4

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web molecular compounds are inorganic compounds that take the form of discrete molecules examples include such familiar substances as water  $\text{H}_2\text{O}$   $\text{H}_2\text{O}$  and carbon dioxide  $\text{CO}_2$   $\text{CO}_2$  these compounds are very different from ionic compounds like

sodium chloride nacl nacl

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web may 28 2020 what is the total mass of hydrogen in each of the molecules a ch<sub>4</sub> b chcl<sub>3</sub> c c<sub>12</sub>h<sub>10</sub>o<sub>6</sub> d ch<sub>3</sub>ch<sub>2</sub>ch<sub>2</sub>ch<sub>3</sub> answer a answer b answer c answer d problem 4 4 1 3 calculate the molecular or formula mass of each of the following a p<sub>4</sub> b h<sub>2</sub>o c ca no<sub>3</sub> d ch<sub>3</sub>co<sub>2</sub>h acetic acid

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