

**Subatomic  
Physics****HOLT PHYSICS****Concept Review***The Nucleus*

1. A certain atom has eight protons, eight electrons, and eight neutrons.

a. How many nucleons does this atom have?

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b. What is the atomic number of this atom?

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c. What is the mass number of this atom?

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d. If the nucleus of this atom has a mass of  $16.124\,552\text{ u}$ , calculate the binding energy of the nucleus.

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e. What is the significance of the binding energy?

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f. Would an atom with eight protons, eight electrons, and nine neutrons be a different element? Explain.

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2. Two protons in a nucleus experience a very large repulsion force.

a. What prevents these two protons from accelerating away from each other?

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b. As a nucleus gets larger, what happens to the ratio of protons to neutrons?

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# Holt Study Guide Subatomic Physics Concept Answers

**Sabine Zange**



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