

Radioactive Isotopes Half-life MCQ Worksheet

- When a radioactive isotope decays over time, its half-life
 - Decreases
 - Increases
 - Stays the same
- As the temperature of a radioactive isotope decreases, its half-life
 - Decreases
 - Increases
 - Stays the same
- A sample of radioactive isotope that initially had a weight of 80 milligrams. It decays to 5 milligrams after 32 days. What is its half-life?
 - 8 days
 - 12 days
 - 4 days
- The half-life of iodine-131 is 8.07 days. How much of a sample of iodine-131 is left after 24.21 days?
 - $\frac{1}{2}$
 - $\frac{1}{3}$
 - $\frac{1}{8}$
- What is the ratio of uranium-238 to lead-206 in a mineral used to determine?
 - Age
 - Density
 - Solubility
- What is the isotope carbon-14 used for?
 - Controlling fission reactions
 - Determining the age of the sample
 - controlling speeds of neutrons
- An isotope has a half-life of 2.5 minutes. What percentage of the isotope is left after 10 minutes?
 - 50 %
 - 3.125 %
 - 6.25 %
- How many milligrams of tritium will remain after 49.2 years if the starting amount is 32 mg? The half-life of tritium is 12.3 years.
 - Age
 - Density
 - Solubility

Half Life Of Radioactive Isotopes Worksheet Answers

Charles H. Perry



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Explained , The steady rate at which unstable isotopes decay known as their half life makes them ideal for dating objects Identify the radioactive isotopes best suited for establishing age such as carbon 14 for organic remains from human history and uranium 238 for billion year old geological formations Also see how stable isotopes can be used for fraud detection and studying ancient climates

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