

Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 3

Section 3.1 Properties of Matter

Label each property as either *physical* or *chemical*.

1. Chemical formula H_2O
2. Forms green carbonate when exposed to moist air
3. Remains unchanged when in the presence of nitrogen
4. Colorless
5. Solid at normal temperatures and pressures
6. Ability to combine with another substance
7. Melting point
8. Liquid at normal temperatures and pressures
9. Boiling point is 100°C
10. Conducts electricity
11. Density is 1.0 g/mL

For each statement below, write *true* or *false*.

12. All matter that we encounter in everyday life exists in one of three physical forms.
13. A solid has definite shape and volume.
14. A liquid has a definite shape and takes on the volume of its container.
15. A gas has both the shape and the volume of its container.
16. The particles in a gas cannot be compressed into a smaller volume.
17. Liquids tend to contract when heated.
18. The particles in a solid are packed far apart.
19. The words *gas* and *vapor* can be used interchangeably.

Section 3.2

What kinds of changes do these words indicate? Write each word under the correct heading. Use each word only once.

	crumble	crush	explode	burn	ferment
Booze					
corrode	rust	laminar	vaporize	oxidize	rot
Physical Change					
Chemical Change					

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____
8. _____

For each item in Column A, write the letter of the matching item in Column B.

Column A

Column B

- | | |
|---|----|
| 17. The new substances that are formed in a chemical reaction | a. |
| chemical change | |
| 18. A chemical reaction that involves one or more substances | b. |
| reactants | |
| changing into new substances | c. |
| products | |

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1. Understanding the eBook Glencoe Chemistry Matter And Change Study Guide Answer Key
 - The Rise of Digital Reading Glencoe Chemistry Matter And Change Study Guide Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Glencoe Chemistry Matter And Change Study Guide Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Glencoe Chemistry Matter And Change Study Guide Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Glencoe Chemistry Matter And Change Study Guide Answer Key
 - Personalized Recommendations
 - Glencoe Chemistry Matter And Change Study Guide Answer Key User Reviews and Ratings

- Glencoe Chemistry Matter And Change Study Guide Answer Key and Bestseller Lists
- 5. Accessing Glencoe Chemistry Matter And Change Study Guide Answer Key Free and Paid eBooks
 - Glencoe Chemistry Matter And Change Study Guide Answer Key Public Domain eBooks
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 - Glencoe Chemistry Matter And Change Study Guide Answer Key Budget-Friendly Options
- 6. Navigating Glencoe Chemistry Matter And Change Study Guide Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Glencoe Chemistry Matter And Change Study Guide Answer Key Compatibility with Devices
 - Glencoe Chemistry Matter And Change Study Guide Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Glencoe Chemistry Matter And Change Study Guide Answer Key
 - Highlighting and Note-Taking Glencoe Chemistry Matter And Change Study Guide Answer Key
 - Interactive Elements Glencoe Chemistry Matter And Change Study Guide Answer Key
- 8. Staying Engaged with Glencoe Chemistry Matter And Change Study Guide Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Glencoe Chemistry Matter And Change Study Guide Answer Key
- 9. Balancing eBooks and Physical Books Glencoe Chemistry Matter And Change Study Guide Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Glencoe Chemistry Matter And Change Study Guide Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Glencoe Chemistry Matter And Change Study Guide Answer Key
 - Setting Reading Goals Glencoe Chemistry Matter And Change Study Guide Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Glencoe Chemistry Matter And Change Study Guide Answer Key
 - Fact-Checking eBook Content of Glencoe Chemistry Matter And Change Study Guide Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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