

Biology Test- Chapter 11: Introduction to Genetics

Multiple Choice

Identify the choice that best completes the statement or answers the question.

1. Gregor Mendel used pea plants to study
 - a. flowering.
 - b. gamete formation.
 - c. the inheritance of traits.
 - d. cross-pollination.
2. Offspring that result from crosses between true-breeding parents with different traits
 - a. are true-breeding.
 - b. make up the F₂ generation.
 - c. make up the parental generation.
 - d. are called hybrids.
3. The chemical factors that determine traits are called
 - a. alleles.
 - b. traits.
 - c. genes.
 - d. characters.
4. Gregor Mendel concluded that traits are
 - a. not inherited by offspring.
 - b. inherited through the passing of factors from parents to offspring.
 - c. determined by dominant factors only.
 - d. determined by recessive factors only.
5. When Gregor Mendel crossed a tall plant with a short plant, the F₁ plants inherited
 - a. an allele for tallness from each parent.
 - b. an allele for tallness from the tall parent and an allele for shortness from the short parent.
 - c. an allele for shortness from each parent.
 - d. an allele from only the tall parent.
6. The principle of dominance states that
 - a. all alleles are dominant.
 - b. all alleles are recessive.
 - c. some alleles are dominant and others are recessive.
 - d. alleles are neither dominant nor recessive.
7. When Gregor Mendel crossed true-breeding tall plants with true-breeding short plants, all the offspring were tall because
 - a. the allele for tall plants is recessive.
 - b. the allele for short plants is dominant.
 - c. the allele for tall plants is dominant.
 - d. they were true-breeding like their parents.
8. A tall plant is crossed with a short plant. If the tall F₁ pea plants are allowed to self-pollinate,
 - a. the offspring will be of medium height.
 - b. all of the offspring will be tall.
 - c. all of the offspring will be short.

Mastering Chemistry Chapter 7 Answers

Judy Breck

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A Problem-Solving Approach to Aquatic Chemistry James N. Jensen, 2023-01-05 A Problem Solving Approach to Aquatic Chemistry Enables civil and environmental engineers to understand the theory and application of aquatic equilibrium chemistry The second edition of A Problem Solving Approach to Aquatic Chemistry provides a detailed introduction to aquatic equilibrium chemistry calculation methods for systems at equilibrium applications of aquatic chemistry and chemical kinetics The text directly addresses two required ABET program outcomes in environmental engineering chemistry including stoichiometry equilibrium and kinetics and material and energy balances fate and transport of substances in and between air water and soil phases The book is very student centered with each chapter beginning with an introduction and ending with a summary that reviews the chapter's main points To aid in reader comprehension important terms are defined in context and key ideas are summarized Many thought provoking discussion questions worked examples and end of chapter problems are also included Each part of the text begins with a case study a portion of which is addressed in each subsequent chapter illustrating the principles of that chapter In addition each chapter has an Historical Note exploring connections with the people and cultures connected to topics in the text A Problem Solving Approach to Aquatic Chemistry includes Fundamental concepts such as concentration units thermodynamic basis of equilibrium and manipulating equilibria Solutions of chemical equilibrium problems including setting up the problems and algebraic graphical and computer solution techniques Acid base equilibria including the concepts of acids and bases titrations and alkalinity and acidity Complexation including metals ligands equilibrium calculations with complexes and applications of complexation chemistry Oxidation reduction equilibria including equilibrium calculations graphical approaches and applications Gas liquid and solid liquid equilibrium with expanded coverage of the effects of global climate change Other topics including chemical kinetics of aquatic systems surface chemistry and integrative case studies For advanced senior undergraduates and first year graduate students in environmental engineering courses A Problem Solving Approach to Aquatic Chemistry serves as an invaluable learning resource on the topic with a variety of helpful learning elements included throughout to ensure information retention and the ability to apply covered concepts in practical settings

Chemistry John S. Phillips, Cheryl Wistrom, 2000

General Chemistry for Engineers Jeffrey Gaffney, Nancy Marley, 2017-11-13 General Chemistry for Engineers explores the key areas of chemistry needed for engineers This book develops material from the basics to more advanced areas in a systematic fashion As the material is presented case studies relevant to engineering are included that demonstrate the strong link between chemistry and the various areas of engineering Serves as a unique chemistry reference source for professional engineers Provides the chemistry principles required by various engineering disciplines Begins with an atoms first approach building from the simple to the more complex chemical concepts Includes engineering case studies connecting chemical principles to solving actual engineering problems Links chemistry to contemporary issues related to the interface between

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Connectivity, the Answer to Ending Ignorance and Separation Judy Breck, 2004 This book contains parallel discussions of how network connectivity is fundamentally diminishing terrorism transforming business enterprises becoming a new artistic expressive medium and providing a new and different locus for human knowledge Connectivity the Answer to Ending Ignorance and Separation is written for every educator eager to learn more about networks

BOOK JACKET Student's Guide, Chemistry, the Central Science James C. Hill, 1991 *Student's Guide to Brown and LeMay, Chemistry, the Central Science, 2nd Edition* James C. Hill, 1981 New York State Regents Exam Kaplan, Kaplan Educational Center Staff, 2000-02

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