

Interpreting the Periodic Table

Examine the hypothetical periodic table below and answer the given questions.

A 10x10 grid with various shaded regions and labels. The grid is composed of 100 squares. Shaded regions include a 2x3 rectangle on the far left, a 3x2 rectangle at the top center, a 2x2 square in the middle center, a 2x2 square on the right side, and a 2x2 square at the bottom right. Labels are placed in specific squares: 'A' is in the bottom-left square, 'B' is in the square below 'A', 'C' is in the square below 'B', 'D' is in the square below 'C', 'E' is in the square below 'D', 'F' is in the square below 'E', 'G' is in the square below 'F', 'H' is in the square below 'G', 'I' is in the square below 'H', 'K' is in the square to the left of the top-left shaded rectangle, 'L' is in the square to the right of the top-center shaded rectangle, 'P' is in the square to the left of the bottom-right shaded rectangle, and 'Z' is in the square to the right of the bottom-right shaded rectangle.

① Which pair of elements has the same number of valence electrons?

② Which pair of metals are in the same period? _____

③ Which element has the smallest atomic number? _____

④ Which elements are metals? _____

⑤ Which elements are nonmetals? _____

⑥ If the atomic number of the element P is 41, what is the atomic number of the element P?

Exam Questions Introduction To Periodic Table

Thomas Griffiths

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