

STANZA BOPAPE SECONDARY SCHOOL**GRADE 11 SCOPE FOR PHYSICAL SCIENCES****JUNE EXAMINATIONS****PAPER 1: PHYSICS****DURATION: 2 HOURS****MARKS: 100****TERMINOLOGY**

Vector, Scalar, resultant vector, normal force, frictional force, inertia, newton 1st, 2nd and 3rd laws, Newton's law of universal gravitation, Coulomb's law, Electric field, electric field at a point, ohm's law, Emt, Current strength, power

QUESTION 1 8 multiple choice questions All topics covered in this scope	[16]
QUESTION 2 <ul style="list-style-type: none"> Vectors in 2D questions. Solving problems involving horizontal and vertical components; resultant vectors, closed vector diagrams 	[10]
QUESTION 3 <ul style="list-style-type: none"> Questions on Newton's law of gravitation Solve problems on gravitational forces, weight and gravitational acceleration 	[8]
QUESTION 4 <ul style="list-style-type: none"> Questions on single body systems on the flat surface, incline plane or hanging, with or without friction. Drawing free-body diagrams Solving problems by applying Newton's 1st, 2nd or 3rd law 	[16]

QUESTION 5 <ul style="list-style-type: none"> Questions on two body systems (joined with a string) on the flat surface, incline plane or hanging, with or without friction Drawing free-body diagrams Solving problems by applying Newton's 1st, 2nd or 3rd law Questions on investigating relationships between acceleration, net force and mass given data tables and/or graphs 	[16]
QUESTION 6 <ul style="list-style-type: none"> Questions on electrostatic forces and electric field between charges Electric field patterns Solving problems involving electrostatic forces and electric field for two or more charges 	[14]
QUESTION 7 <ul style="list-style-type: none"> Questions on Ohm's law experiment including ohmic and non-ohmic conductors Solving circuit problems with a maximum of four resistors connected in series and parallel combinations Explain differences in brightness of light bulbs 	[18]
TOTAL:	100

PAPER 2: CHEMISTRY**DURATION: 2 HOURS****MARKS: 100****TERMINOLOGY**

Covalent bonding, molecules, valence electrons, dative/coordinate covalent bond, electronegativity, bond energy, bond length, interatomic and intermolecular forces, boiling and melting points, vapour pressure, mole, one mole, molar mass, concentration

QUESTION 1 8 multiple choice questions All topics covered in this scope	[16]
QUESTION 2 <ul style="list-style-type: none"> Draw Lewis dot diagrams for atoms, molecules and ions. Answer questions on molecular shapes, polarity of bonds 	[16]
QUESTION 3 <ul style="list-style-type: none"> Determine bond polarity and polarity of molecules by using electronegativities Answer questions on bond energies and bond length given data tables and graphs. 	[14]
QUESTION 4 <ul style="list-style-type: none"> Identify the type of intermolecular forces given the names or formulae of molecules Explain differences in boiling points, melting points and vapour pressures by referring to the types of intermolecular forces and/or molecular masses 	[18]

QUESTION 5 <ul style="list-style-type: none"> Calculations based on molar mass, volume of gas at STP, concentration of aqueous solutions, number of moles, number of atoms/ molecules/ ions etc. Balance chemical equations 	[16]
QUESTION 6 <ul style="list-style-type: none"> Determine percentage composition of compounds, empirical and molecular formulae Solve problems by performing stoichiometric calculations based on balanced equations which include percentage yield. 	[20]
TOTAL:	100

Exam Paper2 2014 June Grade Scope

J Elliott



Exam Paper2 2014 June Grade Scope:

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