

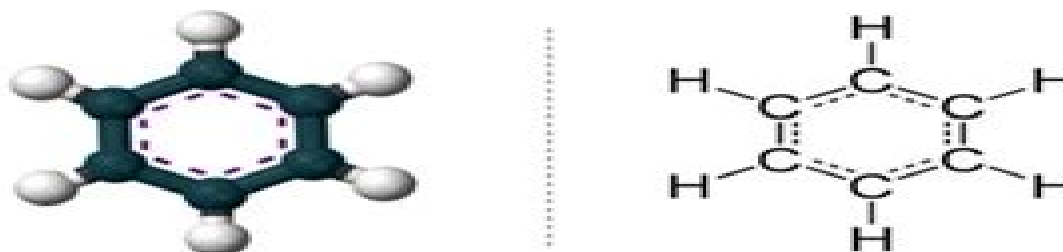
## Hydrocarbons: Alkanes, Alkenes, Alkynes

### Subtopic:

Structure of Hydrocarbons  
Classification and Types of Hydrocarbons  
Properties of Hydrocarbons  
Group of Hydrocarbons  
Uses of Hydrocarbons

### What are Hydrocarbons?

- ✓ Hydrocarbons are *organic compounds* that contains of only two kinds of atoms – *carbon and hydrogen*.
- ✓ Hydrocarbons are *colorless gases* that have very *weak odors*.
- ✓ Hydrocarbons can feature simple or relatively complex structures and can be generally grouped into families, namely *alkanes, alkenes, alkynes, and aromatic hydrocarbons*.
- ✓ Hydrocarbons such as *methane, ethane and butane* are components of *natural gas*. Hydrocarbons such as *propane and butane* are used for commercial fuel purposes in the form of *Liquefied Petroleum Gas (LPG)*. *Benzene*, one of the *simplest aromatic hydrocarbons*, serves as raw material for the synthesis of many *synthetic drugs*.



Structure of Hydrocarbon

- ✓ The *molecular formula* for these compounds is  $C_6H_6$ . The existence of *hydrocarbons is seen in plants and trees*. For example, *Carotenes* is an organic pigment which is found in green leaves and carrots. These hydrocarbons make up to 98% of natural crude rubber. Further, they possess large internal energy which renders them their importance.

### Classification and Types of Hydrocarbons

Older chemists classified hydrocarbons as either *aliphatic* or *aromatic*. The classification was done *based on their source and properties*.

- **Aliphatic hydrocarbons** were derived from *chemical degradation of fats or oils*
- **Aromatic hydrocarbons** contained substances that were a result of *chemical degradation of certain plant extracts*.

# Hydrocarbons Alkanes Study Guide Answers

**Paul A. Bartlett, Judith G. Koch**



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