

Recap/Remember

- 1. A binary of two \_\_\_\_\_ multi \_\_\_\_\_ relationships is a \_\_\_\_\_ structure.
- 2. A binary of \_\_\_\_\_ balanced \_\_\_\_\_ equations

2.  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

2. Magnesium reacting with oxygen to form magnesium oxide. Write the balanced equation.

The \_\_\_\_\_ is a \_\_\_\_\_ reaction. Give the \_\_\_\_\_ ratio \_\_\_\_\_ for the \_\_\_\_\_ involved in the reaction.

E.g. Problem:

When aluminium reacts with oxygen to form aluminium oxide, the reaction is given as:



Write ratios:

Al	Al	O	O
4	Al	2	Al <sub>2</sub> O <sub>3</sub>
3	O <sub>2</sub>	2	Al <sub>2</sub> O <sub>3</sub>

2. If you start with 4 moles of Al, how many moles of Al<sub>2</sub>O<sub>3</sub> will be produced?

4 moles Al gives 2 moles Al<sub>2</sub>O<sub>3</sub> and 3 moles O<sub>2</sub> is used to do

# Episode 801 Note Taking Guide Key

**Christian Drosten**



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