

(iii) What does $\cdot xH_2O$ represent in the formula $Al_2(SO_4)_3 \cdot xH_2O$?

Water of crystallisation

[1]



H_2O crystals to constant mass.

has a mass of 0.046 g

value of x

$$\begin{aligned}H_2O \\ 5.76g \\ \hline 18 \\ = \frac{0.32}{0.02} \\ = 16\end{aligned}$$

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Ying Liu

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