

Empirical and Molecular Formula Worksheet ANSWER KEY

Write the empirical formula for the following compounds.

- 1) C_6H_6 **CH**
- 6) C_8H_{14} **C_4H_7**
- 7) WO_2 **WO_2**
- 8) $C_2H_6O_2$ **CH_3O**
- 9) $X_{10}Y_{12}$ **X_5Y_6**
- 6) A compound with an empirical formula of C_2OH_4 and a molar mass of 88 grams per mole. What is the molecular formula of this compound? **$C_8O_4H_{16}$**
- 7) A compound with an empirical formula of C_4H_4O and a molar mass of 136 grams per mole. What is the molecular formula of this compound? **$C_8H_8O_2$**
- 8) A compound with an empirical formula of $CFBrO$ and a molar mass of 253.8 grams per mole. What is the molecular formula of this compound? **$C_2F_2Br_2O_2$**
- 9) A compound with an empirical formula of C_3H_3N and a molar mass of 46 grams per mole. What is the molecular formula of this compound? **$C_6H_6N_2$**
- 10) A well-known reagent in analytical chemistry, dimethylglyoxime, has the empirical formula C_2H_5NO . If its molar mass is 116.1 g/mol, what is the molecular formula of the compound? **$C_8H_{10}N_2O_2$**
12. A certain blue solid contains 36.84% N and 63.16% O. What is the empirical formula of this compound? **The ratios are $N_{1.50}O_{1.50}$. Since 1.50 is not close to a whole number, we multiply *both* subscripts by 2. The empirical formula is thus N_2O_2 . (The name is dinitrogen trioxide.)**
13. A sample of indium chloride weighing 0.5000 g is found to contain 0.2404 g of chlorine. What is the empirical formula of the indium compound? **$InCl_2$**
14. An unknown compound was found to have a percent composition as follows: 47.0 % potassium, 14.5 % carbon, and 38.5 % oxygen. What is its empirical formula? If the true molar mass of the compound is 166.22 g/mol, what is its molecular formula? **$K_2C_3O_4$**
15. Rubbing alcohol was found to contain 60.0 % carbon, 13.4 % hydrogen, and the remaining mass was due to oxygen. What is the empirical formula of rubbing alcohol? **C_3H_8O**

Empirical Formula Answer Key

Disha Experts

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This book delves into Empirical Formula Answer Key. Empirical Formula Answer Key is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Empirical Formula Answer Key, encompassing both the fundamentals and more intricate discussions.

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- Chapter 1: Introduction to Empirical Formula Answer Key
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- Chapter 3: Empirical Formula Answer Key in Everyday Life
- Chapter 4: Empirical Formula Answer Key in Specific Contexts
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2. In chapter 1, this book will provide an overview of Empirical Formula Answer Key. This chapter will explore what Empirical Formula Answer Key is, why Empirical Formula Answer Key is vital, and how to effectively learn about Empirical Formula Answer Key.

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4. In chapter 3, this book will examine the practical applications of Empirical Formula Answer Key in daily life. This chapter will showcase real-world examples of how Empirical Formula Answer Key can be effectively utilized in everyday scenarios.

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6. In chapter 5, the author will draw a conclusion about Empirical Formula Answer Key. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Empirical Formula Answer Key.

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