

(b) Lithium, carbon (in the form of diamond) and fluorine have very different melting points.

These differences in melting points are the result of different types of structure and different forces or bonds between the particles in the structures.

Part of the table below has been filled in.

Complete the table below.

	Lithium	Carbon (diamond)	Fluorine
Melting point/°C	181	3550	–220
Structure	Giant		Simple
Force or bond overcome on melting	Metallic bond		
Particles between which the force or bond is acting			

[6]

[Total: 10]

END OF QUESTION PAPER

F215 Jun 13 Ocr Question Paper

JA Banks

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