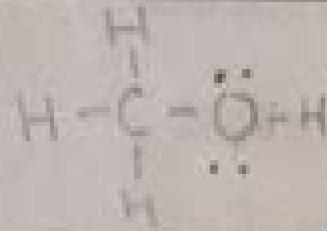
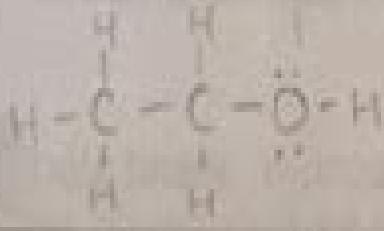
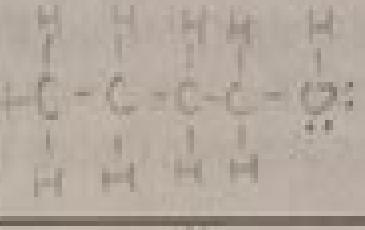
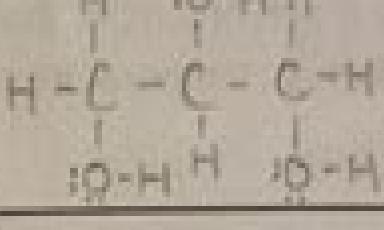
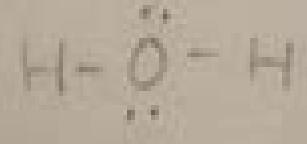


Substance	Lewis Structure	Molar Mass	Intermolecular Attractive Force
Methanol CH_3OH $4 + 3(1) + 16 = 34$		32.042	Hydrogen Bonds
Ethanol $\text{CH}_3\text{CH}_2\text{OH}$ $4 + 5 + 4 + 16 + 1 = 36$		46.062	Hydrogen Bonds
n-Butanol $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$		74.122	Hydrogen Bonds
Glycerin $\text{C}_3\text{H}_{10}(\text{OH})_3$ $12 + 10 + 3 \times 16 = 76$		92.093	Hydrogen Bonds
Water H_2O		18.015	Hydrogen Bonds

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Lauren Gardner

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