

Naming Chemical Compounds Worksheet

Name the following *ionic* compounds:

- 1) NaBr Sodium Bromide
- 2) CaO Calcium Oxide
- 3) Li_2S Lithium Sulfide
- 4) MgBr_2 _____
- 5) $\text{Pu}(\text{OH})_3$ _____
- 6) $\text{Hg}(\text{CN})_2$ _____
- 7) $\text{Mo}(\text{C}_2\text{H}_3\text{O}_2)$ _____
- 8) $\text{Cr}(\text{CrO}_4)_3$ _____
- 9) WO_2 _____
- 10) $\text{Mg}(\text{ClO})_2$ _____

Write the formulas for the following *ionic* compounds:

11. potassium dichromate _____
12. gold(III) oxide _____
13. aluminum periodate _____
14. barium nitrite _____
15. silver carbonate _____
16. lithium sulfite _____
17. zinc hydrogen carbonate _____
18. iron(III) hydroxide _____
19. ammonium phosphate _____
20. copper(II) bromite _____
- 21) SO_3 _____
- 22) N_2S _____
- 23) BF_3 _____
- 24) P_2Br_4 _____
- 25) SiO_2 _____
- 26) SF_6 _____
- 27) NO_2 _____
- 28) nitrogen trichloride _____
- 29) dinitrogen trioxide _____
- 30) phosphorus pentafluoride _____
- 31) diboron tetrahydride _____

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Michael Rajnik



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