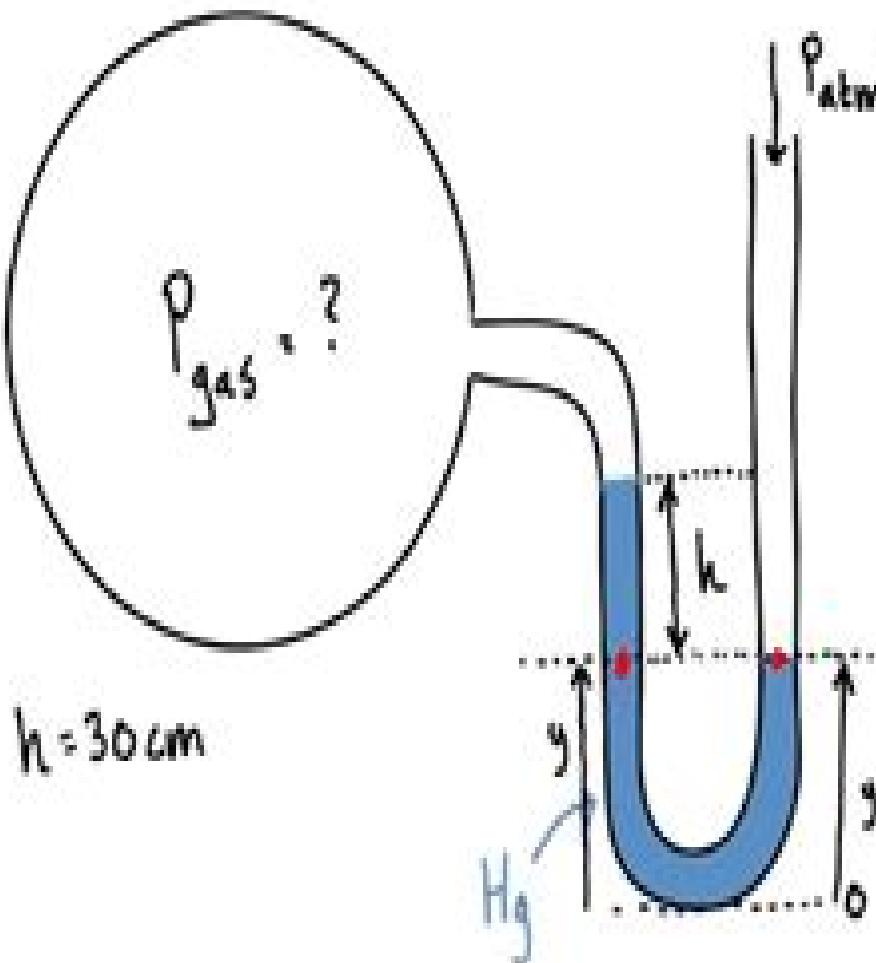


Manometer Example



$$P_{atm} = 101 \text{ kPa}$$

$$P_{gas} = P_{atm} + \rho gh$$

$$= 101 \text{ kPa} - (13,550 \frac{\text{N}}{\text{m}^2})(9.81 \frac{\text{m}}{\text{s}^2})(0.3 \text{ m})$$

$$= 101 \text{ kPa} - 19.9 \text{ kPa}$$

$$P_{gas} = 81.1 \text{ kPa}$$

Manometer Solution Problem

Mike Jess

Manometer Solution Problem:

Chemical Engineering: Solutions to the Problems in Volume 1 J R Backhurst, J H Harker, J.F. Richardson, 2013-10-22 This volume in the Coulson and Richardson series in chemical engineering contains full worked solutions to the problems posed in volume 1 Whilst the main volume contains illustrative worked examples throughout the text this book contains answers to the more challenging questions posed at the end of each chapter of the main text These questions are of both a standard and non standard nature and so will prove to be of interest to both academic staff teaching courses in this area and to the keen student Chemical engineers in industry who are looking for a standard solution to a real life problem will also find the book of considerable interest An invaluable source of information for the student studying the material contained in Chemical

Engineering Volume 1 A helpful method of learning answers are explained in full *Problems and Solutions on Thermodynamics and Statistical Mechanics* Yung-kuo Lim, 1990 Volume 5 *Six-minute Solutions for Civil PE Exam Problems* R. W. Schneiter, 2003

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while the procedures that compliment a specific protocol represent the numerous detailed steps for implementing that protocol This book features over 300 major peer reviewed protocols and nursing procedures on a wide range of clinical topics Each procedure is described in sequential step by step detail Also included are over 300 illustrations checklists tables and flowcharts Icons are used throughout the book to draw attention to the evidence base for a particular protocol alerts for specific considerations related to the protocol and troubleshooting tips for equipment used in the protocol

Principles of

Chemical Engineering Processes Nayef Ghasem, Redhouane Henda, 2014-11-10 This book introduces the basic principles and calculation techniques used in chemical engineering It discusses problems in material and energy balances related to chemical reactors explains the concepts of dimensions units psychrometry steam properties and conservation of mass and energy and demonstrates how MATLAB and Simulink can be used to solve complicated problems This Second Edition contains additional homework problems and a new chapter related to single and multiphase systems Educational software downloadable exercises and a solutions manual are available with qualifying course adoption

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