



Introduction to working as a Chemist.

Use a highlighter as you work through this booklet. ESL students, make vocab lists of the terminology that is defined or described here.

Assessment Statements

- | | |
|--|--|
| <ul style="list-style-type: none"> <input type="checkbox"/> Outline the Scientific Method <input type="checkbox"/> Identify independent, dependent & controlled variables. <input type="checkbox"/> Demonstrate correct and safe use of laboratory equipment. <input type="checkbox"/> Distinguish between quantitative and qualitative data <ul style="list-style-type: none"> ➢ <i>Qualitative is descriptive, not using numbers, e.g. properties, colour, gas production, sound, odor.</i> ➢ <i>Quantitative is mathematical, including</i> <ul style="list-style-type: none"> ◊ <i>Raw data (as recorded)</i> ◊ <i>Processed (after calculations)</i> <input type="checkbox"/> Distinguish between accuracy and precision <ul style="list-style-type: none"> ➢ <i>Accuracy describes the correctness of a set of measurements</i> ➢ <i>Precision describes the repeatability of a set of measurements</i> | <ul style="list-style-type: none"> <input type="checkbox"/> State SI units of measurement for <ul style="list-style-type: none"> ➢ <i>Mass: grams (g)</i> ➢ <i>Length: Metres (m), millimeters (mm)</i> ➢ <i>Concentration: molarity (M)</i> ➢ <i>Temperature: Degrees Celsius (°C)</i> ➢ <i>Volume: Litres (L) or millilitres (mL)</i> ➢ <i>Density: grams per cubic centimeter (g/cm³)</i> ➢ <i>Conductivity: Siemens (S)</i> ➢ <i>Acidity or alkalinity (pH)</i> <input type="checkbox"/> Energy: Joules (J or kJ)Determine the uncertainty of digital and analog measuring tools <ul style="list-style-type: none"> ➢ <i>Usually ± 1 of the smallest division</i> <input type="checkbox"/> Calculate the % error of a given or recorded set of values <ul style="list-style-type: none"> ➢ <i>$(\text{Measured} - \text{True}) / \text{True} \times 100$</i> <input type="checkbox"/> State values to appropriate numbers of significant figures <input type="checkbox"/> Calculate values to appropriate numbers of significant figures |
|--|--|

Related Concepts:

- ☐ 01.1 Element symbols (Quia quiz: <http://www.quia.com/quiz/3246640.htm#>)
- ☐ 01.2 Accuracy, Precision, Error, Uncertainty (Quia quiz: <http://www.quia.com/quiz/3507626.htm#>)
- ☐ 01.3 SigDigs and Notation (Quia quiz: <http://www.quia.com/quiz/3504378.htm#>)

Command Terms: Be sure you understand what is being asked of you

Analyse Break down to identify key parts. Interpret information to reach conclusions.

Apply	Use knowledge & understanding in response to a real situation or given circumstances.
Calculate	Obtain a numerical answer, showing relevant stages in the working.
Define	Give the precise meaning of a word, phrase, concept or physical quantity.
Describe	Give a detailed account or picture of an event, pattern, process or pattern.
Design	Produce a plan, simulation or model.
Determine	Obtain the only possible answer.
Discuss	Offer a considered and balanced review that includes a range of arguments, factors or hypotheses. Opinions or conclusions should be presented clearly and supported by appropriate evidence.
Distinguish	Make clear the differences between two or more concepts or items.
Evaluate	Assess the implications and limitations; make judgments about the ideas, works, solutions or ideas in relation to selected criteria.
Explain	Give a detailed account including reasons or causes.
Identify	Provide an answers from a number of possibilities. Recognise and state briefly a distinguishing fact or feature.
Label	Add title, labels or brief explanation(s) to a diagram or graph.
List	Give a sequence of answers with no explanation.
Outline	Give a brief account.
Recall	Remember or recognize from prior learning experiences.
State	Give a specific name, value or other brief answer without explanation or calculation.
Suggest/Propose	Propose a solution, hypothesis or other possible answer.

Intro Chem Lab Manual Answers

**Uno Kask, Otto W. Nitz, Martin G.
Ondrus, Tammy J. Melton**



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A Laboratory Manual of Organic Chemistry for Beginners Arnold Frederick Holleman,1913 **Chemistry** John Kenkel,Paul B. Kelter,David S. Hage,2000-09-21 What a great idea an introductory chemistry text that connects students to the workplace of practicing chemists and chemical technicians Tying chemistry fundamentals to the reality of industrial life Chemistry An Industry Based Introduction with CD ROM covers all the basic principles of chemistry including formulas and names chemical bon **Study Guide with Selected Solutions** Karen Timberlake,2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems **The Bookseller, Newsdealer and Stationer** ,1899 **Resources in Education** ,1980 *American Horticultural Manual: Comprising the leading principles and practices connected with the propagation, culture, and improvement of fruits, nuts, ornamental trees, shrubs, and plants in the United States and Canada* Joseph L. Budd,Niels Ebbesen Hansen,1902 *The Journal of Industrial and Engineering Chemistry* ,1921 *Chemistry* Karen Timberlake,2003 Chemistry An Introduction to General Organic and Biological Chemistry now in its eighth edition makes chemistry exciting by showing why important concepts are relevant to the lives and future careers of readers The new design digital images photos Career Focus features and macro to micro art enhance the new edition while it retains the many features that have made this book so successful The writing as always is exceptionally friendly Each section contains sample problems that develop readers critical thinking skills This edition also contains more conceptual problems than ever before and has been redesigned to accomodate new styles of learning and teaching with a wide variety of pedagogical tools Health and Environmental Notes throughout the book highlight topics that are relevant to readers lives and are ideal for classroom discussion Explore Your World activities in each chapter make chemistry exciting relevant and non threatening **Journal of Industrial and Engineering Chemistry** ,1921

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