

**Reminders:**

1. In a neutral atom the number of protons equals the number of electrons.
2. An atom can NEVER gain or lose protons.
3. The number of protons equals the atomic number

NAME: \_\_\_\_\_

# Ion Practice Set

1. What is an ion?
2. What does the number next to the ions signify?

*Complete the following table, using the periodic table in the back of your book.*

	ELEMENT NAME	ION SYMBOL	NUMBER OF PROTONS	NUMBER OF ELECTRONS	NUMBER OF ELECTRONS LOST OR GAINED
ex	Fluorine	$F^-$	9	10	gained one
1			53	54	
2			16		gained two
3	potassium				lost one
4		$Ca^{+2}$			
5			35	36	
6		$Sr^{+2}$			
7		$H^+$			
8			8		gained two
9			12		lost two
10	aluminum			10	
11			34	36	
12		$H^-$			
13	lithium				lost one
14		$Rb^+$			
15			17	18	

# Ion Practice Set Answers

**M Walker**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'M Walker' text.

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