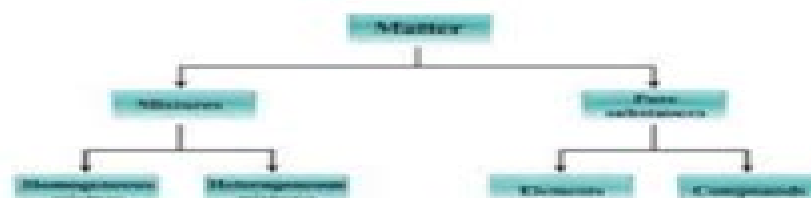


Matter Test Study Guide



matter = anything that has mass and takes up space.

pure substance = a single kind of matter that is pure, meaning that it always has a specific chemical makeup. Ex.) table salt is always NaCl and sugar is always $\text{C}_{12}\text{H}_{22}\text{O}_{11}$

element = a pure substance that can't be broken down into any other substance. Ex.) oxygen, copper, aluminum, zinc, tungsten, carbon, hydrogen, helium, nitrogen

atom = the smallest part of an element. Ex.) one atom of oxygen



compound = a pure substance made of two or more elements chemically combined in a set ratio. Ex.) H_2O

molecule = the smallest part of a compound Ex.) one molecule of H_2O



mixture = made of two or more substances that are mixed together, but not chemically combined. Ex.) trail mix

homogeneous = can't see the different parts. Ex.) sugar water

heterogeneous = can see the different parts. Ex.) salad



solution = an example of a homogenous mixture

chemical formulas = tell you which elements and how many atoms of each are in a compound ex.) $\text{C}_{12}\text{H}_{22}\text{O}_{11}$

set ratio = a specific number of atoms of each element to make a compound

weight = a measure of the force of gravity on an object; can change- depends on location
Ex.) earth vs. moon

mass = a measure of the amount of matter in an object; **does not change**

- triple beam balance
- SI unit is the kilogram
- scientists prefer to measure by mass rather than weight because mass doesn't change



My **WEIGHT** on Earth is around 560N



My **WEIGHT** on the moon is around 90N



My **MASS** is always 56kg!

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G Orfield



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