

TEKS-Aligned STAAR Practice

5. Geometry & Measurement



TEKS
Aligned
ASSESSMENT

Geometry & Measurement Assessment – SET 1

The test assesses mathematics skills
aligned to the TEKS standards.

Geometry & Measurement Assessment – SET 2

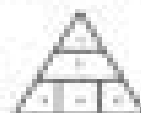
The test assesses mathematics skills
aligned to the TEKS standards.

GRADE 3 Mathematics 10 QUESTION FORMAT

Name _____ Date _____

TEST PRACTICE SET 1 Geometry & Measurement

1. Figure A is composed of 3 identical, smaller figures.



Which figures are identical?

- Figures 1 and 2
- Figures 2 and 3
- Figures 3 and 4
- Figures 1 and 4

2. Figure B is composed of 3 identical, smaller figures. Which figure is not identical to the others?

- 10 identical figures
- 12 identical figures
- 14 identical figures
- 16 identical figures

Name _____ Date _____

TEST PRACTICE SET 2 Geometry & Measurement

1. Figure C is composed of 3 identical, smaller figures. Which figure is not identical to the others?



Which figure is not identical to the others?

- Figure A: 3 identical figures
- Figure B: 4 identical figures
- Figure C: 5 identical figures
- Figure D: 6 identical figures

2. A rectangular prism is composed of 12 identical, smaller figures. Which figure is not identical to the others?

- 10 identical figures
- 12 identical figures
- 14 identical figures
- 16 identical figures

Measuring Up Staar Practice Lesson 16 Key

ML Morrison



Measuring Up Staar Practice Lesson 16 Key:

Resources in Education ,1996-10

Resources in Education ,1996

Sizing Up Measurement Chris Confer,2007

The lessons in Sizing Up Measurement Activities for Grades 3 5 Classrooms focus on length area volume angles weight time and temperature Each lesson is organized in an accessible easy to use format that includes an overview a list of materials a vocabulary list and step by step teaching directions Students come away from these lessons with a deeper understanding of why and how to measure and they develop the confidence required to make sense of any situation and the measurement tools involved pub desc

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transformations can occur as a result of nuclear decay of radioactive isotopes or neutron capture neither procedure requires a change in the atom's element an atom gets transmuted into an atom of a different element when the number of protons in it is changed

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nuclear chemistry is the subfield of chemistry dealing with radioactivity nuclear processes such as nuclear transmutation and nuclear properties it is the chemistry of radioactive elements such as the actinides radium and radon together with the chemistry associated with equipment such as nuclear reactors which are designed to perform

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unlike a normal chemical reaction that forms molecules nuclear reactions result in the transformation of one element into another this property of nuclear reactions is used in nuclear power plants to gather nuclear energy the three common types of radiations are explained below

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a nuclear reactor consists of the following a nuclear fuel a fissionable isotope must be present in large enough quantities to sustain a controlled chain reaction the radioactive isotope is contained in tubes called fuel rods a moderator

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apr 3 2019 in order to understand the origin and character of individual radioactive emissions accompanying nuclear transformation processes we first need to answer three questions what is an unstable nucleus

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in all accelerators the particles move in a vacuum to avoid collisions with gas molecules when neutrons are required for transmutation reactions they are usually obtained from radioactive decay reactions or from various nuclear reactions occurring in nuclear reactors

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nuclear transmutation is the conversion of one chemical element or an isotope into another chemical element 1 nuclear transmutation occurs in any process where the number of protons or neutrons in the nucleus of an atom is changed a transmutation can be achieved either by nuclear reactions in which an outside particle reacts with a nucleus

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jan 10 2021 write a balanced equation for each of the following nuclear reactions bismuth 212 decays into polonium 212 beryllium 8 and a positron are produced by the decay of an unstable nucleus neptunium 239 forms from the reaction of uranium 238 with a neutron and then spontaneously converts into plutonium 239